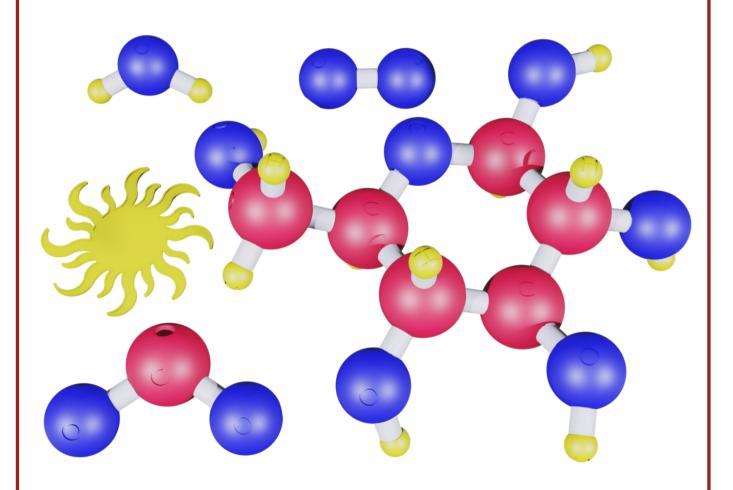
Photosynthesis



Student sheet

Educational resources



www.anhelsnatura.com

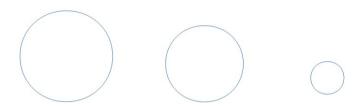
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We place all the pieces on the table and group the like ones.

How many different pieces do we have?

First, we observe that we have 3 types of "balls" that represent the <u>atoms</u> of the different chemical elements that participate in photosynthesis as part of molecules.

• We observe the balls and copy the letter engraved on each one:

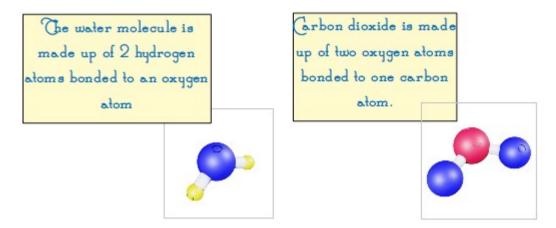


- What do we think these letters mean?
- We find all the chemical elements on the periodic table. We look for the letters on each ball and circle them.

| Group → ↓ Period | • 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
|---------------------|----------|----------|----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|
| 1 | 1 H | | | | | | | | | | | | | | | | | 2 He |
| 2 | 3 Li | 4 Be | | | | | | | | | | | 5 B | 6 C | 7 N | 8 O | 9 F | 10 Ne |
| 3 | 11 Na | 12 Mg | | | | | | | | | | | 13 Al | 14 Si | 15 P | 16 S | 17 Cl | 18 Ar |
| 4 | 19 K | 20 Ca | 21 Sc | 22 Ti | 23 V | 24 Cr | 25 Mn | 26 Fe | 27 Co | 28 Ni | 29 Cu | 30 Zn | 31 Ga | 32 Ge | 33 As | 34 Se | 35 Br | 36 Kr |
| 5 | 37 Rb | 38 Sr | 39 Y | 40 Zr | 41 Nb | 42 Mo | 43 Tc | 44 Ru | 45 Rh | 46 Pd | 47 Ag | 48 Cd | 49 In | 50 Sn | 51 Sb | 52 Te | 53 I | 54 Xe |
| 6 | 55 Cs | 56 Ba | | 72 Hf | 73 Ta | 74 W | 75 Re | 76 Os | 77 Ir | 78 Pt | 79 Au | 80 Hg | 81 TI | 82 Pb | 83 Bi | 84 Po | 85 At | 86 Rn |
| 7 | 87 Fr | 88 Ra | | 104 Rf | 105 Db | 106 Sg | 107 Bh | 108 Hs | 109 Mt | 110 Ds | 111 Rg | 112 Cn | 113 Nh | 114 Fl | 115 Mc | 116 Lv | 117 Ts | 118 Og |
| Lanthanides | | | | 57 La | 58 Ce | 59 Pr | 60 Nd | 61 Pm | 62 Sm | 63 Eu | 64 Gd | 65 Tb | 66 Dy | 67 Ho | 68 Er | 69 Tm | 70 Yb | 71 Lu |
| Actinides | | | | 89 Ac | 90 Th | 91 Pa | 92 U | 93 Np | 94 Pu | 95 Am | 96 Cm | 97 Bk | 98 Cf | 99 Es | 100 Fm | 101 Md | 102 No | 103 Lr |
| | | | | | | | | | | | | | | | | | | |

Using the internet, we find out which chemical element each symbol corresponds to. Now we know what the "balls" are. And the remaining piece? Perhaps we've guessed that it's used to join the "balls." We'll call this other piece a chemical bond, or, to simplify, a bond. What do atoms form when they join together? ____ We notice that there are holes in the balls; how many bonds do we think the carbon atom can form? Why? (Ind oxygen? (Ind hydrogen? O:______H:____ We observe the photosynthesis reaction. To start photosynthesis, the plant needs: water (120), carbon dioxide ((O2), and energy. We indicate with arrows in the drawing where it gets it. H_2O

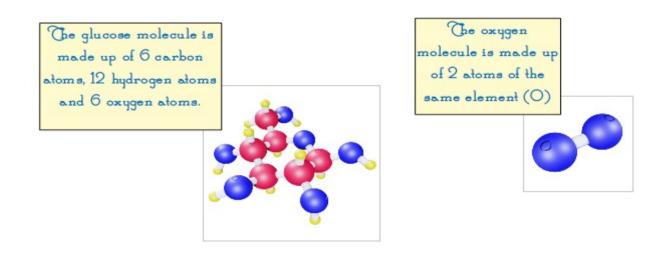
Mext, we assemble the molecules needed to start photosynthesis using the atoms and bonds:



We observe the photosynthesis reaction: how many water molecules do we need? ___ (Ind carbon dioxide? ___ We assemble them all.

$$6H_2O + 6CO_2$$
 $C_6H_{12}O_6 + 6O_2$

Now we disassemble them, break the bonds, and build the molecules that are the product of the reaction: six oxygen molecules (O_2) and one glucose molecule $(C_6 \text{ M}_{12}O_2)$. We observe that we must exert an effort to do this work and that the number of atoms remains constant (no more, no less).



| • Where do we get the energy to do this? |
|---|
| What type of energy is it? Like us, plants need energy |
| to break and form bonds; where do plants get that energy? |
| What type of energy is it? |
| |
| Using all the elements of the product (numbers, arrow, sun), we |
| "assemble" the entire photosynthesis reaction. |
| 6 |
| What importance do we give to this process? |
| |
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